Polarity Matching of Radical Trapping: High Yielding 3-exo and 4-exo **Cyclizations**

Andreas Gansäuer,* Thorsten Lauterbach, and Daniel Geich-Gimbel^[a]

Abstract: A catalyzed synthesis of cyclopropanes and cyclobutanes via radical chemistry is described. The method that generally proceeds in high yields uses epoxides as radical precursors and titanocene(iii) complexes as the electron transfer catalysts (see scheme). The key to the success of the transformation is constituted by the chemoselectivity of radical reduction. Electrophilic enol radicals generated through cyclization are reduced rapidly whereas their precursors, nucleophilic alkyl radicals, remain unaffected.

Introduction

Radical cyclization reactions are amongst the most powerful and versatile methods for the construction of mono- and polycyclic systems.[1] However, the synthesis of small rings, that is, cyclopropanes, a highly interesting functional group under theoretical, preparative and structural aspects, $[2]$ and cyclobutanes, is usually difficult. In the former case this is because ring openings of the intermediate cyclopropyl carbinyl radicals are amongst the fastest radical reactions known $(k \sim 10^{8}-10^{9} s^{-1})^{[3]}$ and in the latter case, ring closure is one of the slowest radical reactions $(k-1 s^{-1})$ while ring opening of the cyclobutylcarbinyl radical is fast,[4] as shown in Scheme 1.

Scheme 1. Kinetics of the 3-exo and 4-exo cyclization.

In this publication we wish to report our first results on high yielding, catalytic, and tin free^[5] radical syntheses of cyclopropanes and cyclobutanes circumventing these prob-

Chem. Eur. J. 2004, 10, 4983-4990 DOI: 10.1002/chem.200400685 © 2004 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim \longrightarrow 4983

Keywords: cyclobutanes · cyclopropanes · epoxides · radical reactions · titanium

lems. The method is complementary to the concepts used in traditional free radical chemistry for the synthesis of small rings. In these cases the stabilization of the carbinyl radicals through conjugation and incorporation of some of the ring strain of the cyclopropanes into the starting materials have been utilized with good success.^[6]

Our approach exploits the different and polarity controlled reactivity of alkyl radicals generated from epoxides^[7,8] and enol radicals generated after cyclization towards [Cp2TiCl], the electron transfer reagent used in this study. This allows both formation and trapping of the unstable carbinyl radicals as shown in Scheme 2.

Electrophilic Radical quickly reduced by Cp₂TiCl

Scheme 2. Polarity matching for radical trapping.

Nucleophilic Radical

slowly reduced by Cp₂TiCl

At first glance our approach may seem similar to the one employed by Fernández-Mateos in his elegant stoichiometric cyclization reactions of epoxide derived radicals to aldehydes and ketones^[7e] or to nitriles.^[7f] However, both reactions do not work under our catalytic conditions and must therefore be considered as conceptually different. While a detailed study on this discrepancy would have to be carried out to establish the differences, we suggest hydrogen atom abstraction from THF by the highly electrophilic and shortlived radicals generated under Fernández-Mateos' conditions as a plausible cause.^[9] Catalyst regeneration would

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then become difficult. This is not the case under our conditions.

In free radical chain reactions polarity controlled transformations have been used with excellent success, for example, in Roberts' polarity reversed catalysis.^[10]

Nucleophilic alkyl radicals generated through epoxide opening are only slowly reduced by $[Cp_2TiC]$. Since no other radical pathway than cyclization is available under our reaction conditions they remain fairly stable. This results in increased longevity that is decisive for the very slow formation of the cyclobutyl carbinyl radicals in the 4-exo cyclization. The unfavorable equilibrium for the formation of both rings is therefore irrelevant if the radicals generated through cyclization are reduced efficiently. The swift and exothermic trapping of the electrophilic enol radical to yield titanium enolates provides this desired electronic differentiation and thus a driving force for irreversible termination of small ring formation.

Results and Discussion

We decided to check the validity of the concept for 3-exo cyclizations first to avoid anticipated problems of the very slow 4-exo cyclization. The results summarized in Table 1 amply demonstrate the validity of our approach.

Table 1. Titanocene catalyzed 3-exo cyclizations.[10] For reasons of clarity only the major isomer is shown.

Entry	Substrate	Product	Yield [%]	may be mono- (11, entry 6), di- (9, 13, 15, entries 5, 7, 8),
$\mathbf{1}$	COMe	COMe $\overline{2}$ OН	72	and trisubstituted $(1, 3, 5,$ entries 1, 2, 3). Thus tetra- $(2, 4, 5, 5, 5, 5, 6)$ 6, and 12, entries $1-3$, 6) and even pentasubstituted cyclo- propanes $(10, 14,$ and 16 , entries 5, 7, and 8) can be easily prepared by our method. Diastereoselectivities are in the
\overline{c}	CO ₂ Et 3	CO ₂ Et	92	usual range for radical cyclizations. The substrates for the cyclization were prepared according to standard procedures from known compounds as shown in Schemes 3 and 4.
3	CONMe ₂ 5	CONMe ₂ 6 ŌН	96	
4	Ph $\overline{7}$	Ph 8 ŌН	$<\!10$	HWE or Wittig olefination 17 3, R= COOEt, 70% 5, R= CONMe ₂ 56% 7, R= Ph, 33%
5	CONMe ₂ 9 nP	10 HO. CONMe ₂ nPr	95, dr 63:37	Wittig-reaction reduction
6	CONMe ₂ ୍ 11	HO. CONMe ₂ 12	88, dr 81:19	19, 44% 18 olefination Swern oxid. `OH ۲O
7	CONMe ₂ 13	-OH CONMe ₂ 14	72, dr 75:25	20, 88% 21, 73% O epoxidation NMe ₂ NMe ₂
8	CONMe ₂ 15	ЮH COMMe ₂ 16	84, dr 75:25	22, 90% 9,92% Scheme 3. Synthesis of 3- <i>exo</i> cyclization precursors. See Experimental Section for details.

With the β -ionone derived epoxides 1, 3, and 5 good to excellent yields of the desired cyclopropanes were achieved through polarity controlled reduction (entries 1–3). Diastereoselectivities of the cyclizations were excellent. It is important to note the good yield of the reaction of 1 containing an enone.^[11] As this sensitive functional group is reduced rapidly by other popular electron transfer reagents, for example, SmI_2 , ^[12] the mildness of the radical generation from epoxides with $[Cp_2TiCl]$ is impressively highlighted.

The alternative approach of a thermodynamic stabilization of the cyclopropyl carbinyl radical, that has been applied with success in traditional free radical chemistry together with incorporation of some of the cyclopropane's strain $[6, 13]$ in the starting material, was also investigated (entry 4). With epoxide 7 cyclization should lead to a highly stabilized benzyl radical. This intermediate is, however, reduced even more slowly by $[Cp_2TiC]$ than simple alkyl radicals.[8l] As none of the desired product could be isolated it is clear that the thermodynamic promotion of the 3-exo cyclization is not relevant for the other examples in table 1 also, since radical stabilization by ketone, ester, and amide groups (entries 1–3) is much smaller compared with arenes,^[14] especially when a tertiary radical adds to the α , β unsaturated carbonyl compounds (all entries except entry 6).

The other cases shown demonstrate that substitution patterns difficult to access with other methods^[15] can be readily obtained in good to excellent yields. The epoxide precursors

Scheme 4. Synthesis of precursors for the 3-exo cyclization. See Experimental Section for details.

We then focussed our attention to 4-exo cyclizations. It turned out that under our conditions excellent results could be obtained as shown in Scheme 5.

Scheme 5. Titanocene catalyzed 4-exo cyclization.

The high yields indicate that the longevity of the alkyl radicals generated by electron transfer from titanocene(III) chloride is sufficient to enable the unusually slow 4-exo cyclization^[16] without interference of competing radical pathways. This results in an almost quantitative trapping of the enol radicals obtained after cyclization to give the cyclobutanes in excellent yield. The starting materials were prepared as shown in Scheme 6.

Efficient 4-exo cyclizations via radical chain processes are rare and require special substrates or stabilization of the products.^[16a,b] In SmI_2 chemistry only cyclization of ketyl radicals to enoates are known. However, at least in the case of ketones it has been unequivocally demonstrated that the

Scheme 6. Synthesis of precursors for the 4-exo cyclization. See Experimental Section for details.

enoate is reduced before the carbonyl group.^[16d] To the best of our knowledge, our examples belong to the most efficient 4-exo cyclizations known as yet.

In summary, the concept of polarity matching of radical trapping has provided a highly efficient and catalytic method for the preparation of cyclopropanes and cyclobutanes through radical chemistry. It also avoids the toxicity problems associated with tin reagents.

Experimental Section

General procedures: All reactions were performed in oven-dried (100°C) glassware under Ar. THF was freshly distilled from K. Et₂O was freshly distilled from Na/K. CH₂Cl₂ was freshly distilled from CaH₂. The products were purified by flash chromatography on Merck silica gel 50 (eluents given in brackets, EE refers to ethyl acetate, CH to cyclohexane, PE to petrol ether 40–60 °C).^[17] Yields refer to analytically pure samples. Isomer ratios were determined by suitable ¹H NMR integrals of cleanly separated signals. NMR: Bruker AMX 300, AM 400; ¹H NMR, [D₅]benzene $(7.16$ ppm) and CHCl₃ $(7.26$ ppm) in the indicated solvent as internal standard in the same solvent; 13 C NMR: CDCl₃ (77.16 ppm) and C_6D_6 (138.06 ppm) as internal standard in the same solvent; integrals in accord with assignments, coupling constants are measured in Hz and always constitutes $J(H,H)$ coupling constants. An asterisk $(*)$ indicates the signal of the minor diastereomer. Combustion analyses: Mr. Frank Hambloch, Institut für Organische Chemie der Georg-August-Universität Göttingen and Mrs. Annie Martens, Kekulé-Institut für Organische Chemie der Rheinischen Friedrich-Wilhelms-Universität Bonn. IR spectra: Perkin-Elmer 1600 series FT-IR as KBr-Pellets or neat films on KBr plates.

The following compounds were prepared according to literature procedures, or have already been described in the literature: The following compounds were synthesized according to literature procedures: 1, [18] $\,17,^{[19]}$ $18,^{[20]}$ $23,^{[21]}$ $25,^{[22]}$ $27,^{[23]}$ $36.^{[24]}$

General procedure 1 (GP 1)

Horner–Wadsworth–Emmons olefination: NaH (96.0 mg 4.00 mmol) was suspended in dry Et₂O (30 mL). The phosphonate (5.00 mmol) was added dropwise to the suspension. After the reaction mixture became clear (usually 5 min except 2 h for the tert-butyl) the aldehyde (3.00 mmol) was added. The solution was stirred at room temperature for the indicated time. The organic phase was washed with water $(2 \times 15 \text{ mL})$, the combined aqueous layers were extracted with CH_2Cl_2 (10 mL). The combined organic layers were washed with brine (5 mL) and dried (MgSO4). The solvents were removed under reduced pressure and the crude product purified by silica gel chromatography.

General procedure 2 (GP 2)

Epoxidation of olefins by mCPBA: The olefin (2.00 mmol) was dissolved in dichloromethane (20 mL). At 0° C mCPBA (518 mg, 3.00 mmol) was added. The reaction mixture was stirred at room temperature for the indicated time. After adding of $Et₂O$ (50 mL) the organic phase was washed with $2N$ NaOH (2×10 mL), the combined aqueous layers were

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extracted with CH_2Cl_2 (10 mL). The combined organic layers were washed with brine (5 mL) and dried $(MgSO₄)$. The solvents were removed under reduced pressure and the crude product purified by silica gel chromatography.

General procedure 3 (GP 3)

Radical cyclization: The epoxide (1.00 mmol) was added to a suspension of $[Cp_2TiCl_2](24.9 \text{ mg}, 100 \mu \text{mol})$, collidine hydrochloride (394 mg, 2.50 mmol) and zinc dust (131 mg, 2.00 mmol), that were heated under vacuum until the collidine hydrochloride began to sublime slightly prior to use, in dry THF (10 mL) and the reaction mixture stirred at room temperature for the indicated time.

After adding of $Et₂O$ (20 mL) the reaction mixture was washed with of $2N$ HCl (2×10 mL), in case of amides with water (2×10 mL), the combined aqueous layers were extracted with CH_2Cl_2 (10 mL). The combined organic layers were washed with brine (5 mL) and dried $(MgSO₄)$. The solvents were removed under reduced pressure and the crude product purified by silica gel chromatography.

1-(2-Hydroxy-1,5,5-trimethyl-bicyclo[4.1.0]hept-7-yl)-propan-2-one (2): According to GP 3 1 (417 mg, 2.00 mmol), $[Cp_2TiCl_2]$ (48.8 mg, 200 µg), collidine hydrochloride (788 mg, 5.00 mmol) and zinc dust (262 mg, 4.00 mmol) for 16 h. Silica gel chromatography (CH/EE 9:1) gave 2 (302 mg, 72%). $R_f = 0.4$ (CH/EE 9:1); ¹H NMR (300 MHz, CDCl₃): $\delta =$ 3.82 (t, $3J=6.1$ Hz, 1H), 3.99 (brs, 1H), 2.80 (dd, $3J=18.9$, $3J=4.2$ Hz, 1H), 2.13 (s, 3H), 2.12 (dd, $\frac{2J}{-19.1}$, $\frac{3J}{-10.3}$ Hz, 1H), 1.57 (dddd, $\frac{2J}{-1}$ 14.1, ${}^{3}J=8.7$, 5.4, 3.3 Hz, 1H), 1.25–1.04 (m, 2H), 1.01 (s, 3H), 1.00 (s, 3H), 0.99–0.88 (m, 2H), 0.87 (s, 3H), 0.40 ppm (d, $3J=5$, 8 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃): $δ = 210.2$, 72.2, 43.1, 39.8, 33.6, 31.6, 29.7, 28.6, 27.2, 19.9, 16.5 ppm; IR (film): $\tilde{v} = 3475$, 2950, 1710, 1465, 1410, 1360, 1305, 1235, 1165, 1070, 1050, 985, 895 cm⁻¹; HRMS (EI, 70 eV): $m/$ z: calcd for $C_{13}H_{22}O_2$: 210.1620; found: 210.1631 [M]⁺.

3-(1,3,3-Trimethyl-7-oxa-bicyclo[4.1.0]hept-2-yl)-acrylic ethyl ester (3): According to GP 1 17 (1.68 g, 10.0 mmol), NaH (300 mg, 12.5 mmol), the phosphonate (3.36 g, 15.0 mmol) in THF (50 mL) for 16 h. Silica gel chromatography (CH/EE 9:1) to yield 3 (1.62 g, 70%). $R_f = 0.4$ (CH/EE 17:3); ¹H NMR (400 MHz, C₆D₆): δ = 7.26 (dd, ³J = 15.7, 10.4 Hz, 1 H), 5.97 (dd, $3J=15.7$, $5J=0.6$ Hz, 1H), 4.03 (qd, $3J=7.1$, $5J=1.5$ Hz, 2H), 2.65 (dd, $3J=2.5$, 1.6 Hz, 1H), 1.76 (d, $3J=10.4$ Hz, 1H), 1.69 (dddd, $3J=10.4$ 15.2, ${}^{3}J=5.3$, 3.9, 1.4 Hz, 1H), 1.43 (dddd, ${}^{2}J=15.2$, ${}^{3}J=10.8$, 5.9, 2.7 Hz, 1H), 1.25 (ddd, $\frac{3}{7}$ = 13.5, $\frac{3}{7}$ = 10.8, 5.3 Hz, 1H), 1.02 (s, 3H), 0.97 (t, $\frac{3}{7}$ = 7.1 Hz, 3H), 0.63 (s, 3H), 0.59 (s, 3H), 0.63–0.55 ppm (m, 1H); 13C NMR $(100 \text{ MHz}, \text{ C}_6\text{D}_6)$: $\delta = 165.6$, 147.0, 124.3, 60.1, 58.8, 58.1, 52.2, 31.1, 28.7, 27.7, 27.2, 23.9, 22.0, 14.3 ppm; IR (film): $\tilde{v} = 2960$, 1720, 1650, 1450, 1365, 1340, 1305, 1260, 1215, 1160, 1115, 1095, 1045, 990, 960, 905 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₄H₂₂O₃: 238.1569; found: 238.1575 $[M]$ ⁺.

(2-Hydroxy-1,5,5-trimethyl-bicyclo[4.1.0]hept-7-yl)-acetic ethyl ester (4): According to GP 3 3 (238 mg, 1.00 mmol), $[Cp_2TiCl_2]$ (24.9 mg, 100 µg), collidine hydrochloride (394 mg, 2.50 mmol) and zinc dust (131 mg, 2.00 mmol) for 16 h. Silica gel chromatography (CH/EE 49:1) gave 4 (222 mg, 92%). $R_f = 0.1$ (CH/EE 19:1); ¹H NMR (400 MHz, C₆D₆): $\delta =$ 3.93–3.88 (m, 1H), 3.91 (q, $3J=7.1$ Hz, 2H), 2.70 (dd, $3J=9.8$, 5.0 Hz, 1H), 2.34 (dd, $\mathrm{^{2}J=17.7}$, $\mathrm{^{3}J=5.1}$ Hz, 1H), 1.81 (dd, $\mathrm{^{2}J=17.6}$, $\mathrm{^{3}J=10.3}$ Hz, 1H), 1.55 (dddd, $^2J=14.1$, $^3J=8.8$, 5.5, 3.3 Hz, 1H), 1.29 (dddd, $^2J=13.9$, $3J=9.5, 6.7, 2.8$ Hz, 1H), 1.11 (ddd, $2J=13.7, 3J=9.0, 2.8$ Hz, 1H), 1.04 (s, 3H), 1.00 (ddd, $\frac{3}{5}J=10.6$, $\frac{3}{5}J=9.5$, 3.3 Hz, 1H), 0.95 (t, $\frac{3}{5}J=7.2$ Hz, 3H), 0.93 (s, 3H), 0.91–0.86 (m, 1H), 0.84 (s, 3H), 0.20 ppm (d, $3J = 5.8$ Hz, 1H); ¹³C NMR (100 MHz, C₆D₆): δ = 174.0, 71.8, 60.5, 39.8, 33.6, 33.6, 31.4, 28.6, 28.2, 27.6, 27.4, 19.8, 17.6, 14.1 ppm; IR (film): $\tilde{v} = 3450$, 2950, 1720, 1465, 1370, 1310, 1195, 1070, 1035, 930, 855 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₄H₂₄O₃: 240.1725; found: 240.1728 [M]⁺.

N,N-Dimethyl-3-(1,3,3-trimethyl-7-oxa-bicyclo[4.1.0]hept-2-yl)-acrylam-

ide (5): According to GP 1 17 (466 mg, 2.77 mmol), NaH (72.0 mg, 3.00 mmol), the phosphonate (781 mg, 3.50 mmol) in THF (25 mL) for 16 h. Silica gel chromatography (CH/EE 1:1) to yield 5 (368 mg, 56%). $R_{\rm f}$ =0.4 (EE); ¹H NMR (400 MHz, C₆D₆): δ =7.12 (d, ³J = 14.5, 10.4 Hz, 1H), 6.15 (d, $\sqrt[3]{J}$ = 15.7 Hz, 1H), 2.72 (brs, 3H), 2.71 (dd, $\sqrt[3]{J}$ = 2.3, 1.1 Hz, 1H), 2.41 (brs, 3H), 1.85 (d, $\delta J = 10.0$ Hz, 1H), 1.75 (dddd, $\delta J = 15.1$, $\delta J =$ 5.3, 4.1, 1.2 Hz, 1H), 1.50 (dddd, $\frac{2}{J}$ =15.1, $\frac{3}{J}$ =10.3, 5.7, 2.8 Hz, 1H), 1.33 $(\text{ddd}, \, \, \, \frac{3}{5}J = 10.3, \, \, 5.3 \, \text{Hz}, \, \, 1 \, \text{H}), \, \, 1.13 \, \text{(s, 3H)}, \, \, 0.73 \, \text{(s, 3H)}, \, \, 0.69 \, \text{(s, 3H)}$ 3H), 0.70–0.61 ppm (m, 1H); ¹³C NMR (100 MHz, C₆D₆): δ = 165.8,

142.9, 123.7, 58.7, 58.2, 52.3, 36.5, 35.1, 31.3, 29.1, 28.1, 27.2, 24.1, 22.1 ppm; IR (film): $\tilde{v} = 2930, 1660, 1610, 1415, 1390, 1305, 1260, 1150,$ 1110, 1045, 990, 895, 870, 805, 745, 715, 650, 620 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₄H₂₃NO₂: 237.1729; found: 137.1734 [M]⁺.

2-(2-Hydroxy-1,5,5-trimethyl-bicyclo[4.1.0]hept-7-yl)-N,N-dimethylacetamide (6): According to GP 3 5 (237 mg, 1.00 mmol), $[Cp_2TiCl_2]$ $(24.9 \text{ mg}, 100 \text{ \mu g})$, collidine hydrochloride $(394 \text{ mg}, 2.50 \text{ mmol})$ and zinc dust (131 mg, 2.00 mmol) for 16 h. Silica gel chromatography (CH/EE 17:5) gave 6 (229 mg, 96%). $R_f = 0.1$ (CH/EE 17:5); ¹H NMR (400 MHz, C_6D_6 : $\delta = 4.78$ (brs, 1H), 4.09 (dd, ³J = 5.8, 5.8 Hz, 1H), 2.56 (s, 3H), 2.20 (dd, \degree J = 17.2, \degree J = 4.2 Hz, 1H), 2.10 (s, 3H), 1.68 (dddd, \degree J = 13.9, $3J=9.6, 5.8, 3.6$ Hz, 1H), 1.51 (dddd, $2J=14.3, 3J=8.4, 6.0, 2.8$ Hz, 1H), 1.42 (dd, \degree J = 17.2, \degree J = 11.2 Hz, 1H), 1.26 (ddd, \degree J = 13.6, \degree J = 9.7, 3.0 Hz, 1 H), 1.12 (ddd, $\mathrm{^{3}J=11.2, 6.4, 4.2 \, Hz, 1 \, H}$), 1.09 (s, 3 H), 1.03 (s, 3 H), 0.96 $(\text{ddd}, \, \, \, \text{J} = 13.9, \, \, \, \, \text{J} = 9.0, \, \, \text{3.4 Hz}, \, \, \text{J} = 11, \, \, \text{O} = 94, \, \, \text{(s)} \, \, \text{J} = 11, \, \, \text{(t)} \, \, \text{J} = 5.9 \, \, \text{Hz}$ 1H); ¹³C NMR (100 MHz, C₆D₆): δ = 172.6, 71.3, 39.4, 35.7, 34.9, 33.3, 32.2, 31.3, 29.0, 28.4, 28.0, 26.9, 20.1, 18.1 ppm; IR (film): $\tilde{v} = 3350$, 2950, 2870, 1610, 1500, 1460, 1415, 1360, 1270, 1150, 1120, 1035, 1000, 920, 855, 795 cm⁻¹; elemental analysis calcd (%) for C₁₄H₂₅NO₂ (239.35): C 70.25, H 10.53, N 5.85; found: C 69.96, H 10.46, N 5.79.

1,3,3-Trimethyl-2-styryl-7-oxa-bicyclo[4.1.0]heptane (7): Benzyltriphenylphosphonium bromide (3.47 g, 8.00 mmol) was suspended in THF (60 mL). At 0° C *n*-butyl lithium (384 mg (2.4 M in hexane), 6.00 mmol) was added dropwise (5 min). After the reaction mixture was stirred for 30 min at 0° C, 17 (1.01 g, 6.00 mmol) was added dropwise (10 min). The solution was allowed to stir at rt for 16 h. After addition of water (30 mL) triphenylphosphine oxide was precipitated by dropwise addition of PE. The organic layer was filtered and dried (MgSO₄). The solvents were removed under reduced pressure and the crude product purified by silica gel chromatography (CH/EE 49:1) to yield 7 (480 mg, 33%, dr 10:1). $R_f = 0.4$ (CH/EE 19:1); ¹H NMR (400 MHz, C₆D₆): $\delta = 7.35 - 7.29$ $(m, 2H), 7.14-7.09$ $(m, 2H), 7.07-7.01$ $(m, 1H), 6.44$ $(dd, {}^{3}J=15.9,$ 9.3 Hz, 1H), 6.36 (d, $3J=15.8$ Hz, 1H), 2.79 (dd, $3J=2.8$, 1.4 Hz, 1H), 1.89 (dd, $3J=9.2$, $4J=0.8$ Hz, 1H), 1.84 (dddd, $2J=15.2$, $3J=5.3$, 4.4, 1.1 Hz, 1H), 1.57 (dddd, $\mathcal{I} = 15.1$, $\mathcal{I} = 10.2$, 5.8, 3.0 Hz, 1H), 1.43 (ddd, 2J =13.2, 3J =10.1, 5.3 Hz, 1H), 1.18 (s, 3H), 0.79 (s, 3H), 0.78 (dddd, 2J = 13.3, $\frac{3}{7}$ = 5.6, 4.5, $\frac{4}{7}$ = 1.0 Hz, 1H), 0.66 ppm (s, 3H); ¹³C NMR (100 MHz, C_6D_6 : δ = 138.0, 133.0, 129.0, 128.8, 127.3, 126.7, 59.2, 59.0, 53.3, 31.6, 29.8, 28.3, 27.2, 24.1, 22.4 ppm; IR (film): $\tilde{v} = 3025$, 2955, 1600, 1940, 1450, 1380, 1305, 1180, 1145, 1095, 1045, 975, 895, 860, 800, 755, 725, 695 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₇H₂₂O: 242.1671; found: 242.1664 $[M]$ ⁺.

2,2-Dimethyl-3-propyl-but-3-enoic ethyl ester (19): NaH (661 mg, 28.8 mmol) was suspended in dry DMSO (30 mL) and heated at 70° C for 1 h. Methyltriphenylphosphonium bromide (10.3 g, 28.8 mmol) in DMSO (30 mL) was added. After 1 h 18 (4.66 g, 25.0 mmol) was added and the reaction mixture was stirred at 55°C for 16 h. The Reaction was stopped by adding water (60 mL) and extracting with Et₂O (3×50 mL). The combined organic layer was washed with water/DMSO (1:1, 50 mL), brine (20 mL) and dried $(MgSO₄)$. The solvents were removed under reduced pressure and the crude product purified by silica gel chromatography (PE/Et₂O 99:1) to yield 19 (2.02 g, 44%). $R_f = 0.4$ (PE/Et₂O 19:1); ¹H NMR (400 MHz, CDCl₃): δ = 4.98 (s, 1H), 4.87 (s, 1H), 4.11 (q, ³J = 7.1 Hz, 2H), 1.93 (t, $3J=7.7$ Hz, 2H), 1.48 (qt, $3J=7.7$, 7.6 Hz, 2H), 1.31 $(s, 6H)$, 1.22 (t, $\frac{3J}{7}$ = 7.1 Hz, 3H), 0.91 ppm (t, $\frac{3J}{7}$ = 7.3 Hz, 3H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: $\delta = 176.7, 152.0, 108.3, 60.5, 48.1, 34.3, 24.8, 21.6, 14.1,$ 14.0 ppm; IR (film): $\tilde{v} = 2960, 1730, 1640, 1465, 1380, 1255, 1135, 1030,$ 895 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₁H₂₀O₂: 184.1463; found: 184.1454 $[M]$ ⁺.

2,2-Dimethyl-3-propyl-but-3-en-1-ol (20): Compound 19 $(8.29 \text{ g},$ 45.0 mmol) was added dropwise to a suspension of $LiAlH₄$ (854 mg, 22.5 mmol) in dry THF at 0° C. The reaction mixture was stirred for 24 h at room temperature. Water (1.5 mL), NaOH (1.5 mL, 15% w/v) and water (4.5 mL) were added subsequently. The solid was filtered off and the solvents were removed under reduced pressure to yield 20 (5.62 g, 88%). ¹H NMR (400 MHz, CDCl₃): δ = 4.94 (s, 2H), 4.38 (s, 2H), 1.97 (t, $3J=7.8$ Hz, 2H), 1.27 (qt, $3J=7.6$, 7.5 Hz, 2H), 1.05 (s, 6H), 0.94 ppm (t, $3J=7.3$ Hz, 3H); ¹³C NMR (100 MHz, C₆D₆): δ = 153.4, 110.0, 69.9, 41.6, 33.5, 24.0, 22.0, 14.1 ppm; IR (film): $\tilde{v} = 3390$, 2960, 1633, 1465, 1040, 895, 825, 640 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₉H₁₈O: 142.1356; found: 142.1356 $[M]$ ⁺.

2,2-Dimethyl-3-propyl-but-3-enal (21): DMSO (6.56 g, 42.0 mmol) in $CH₂Cl₂$ (5 mL) was added dropwise to a solution of oxalyl chloride (5.33 g, 42.0 mmol) in CH_2Cl_2 (150 mL) at -60 °C. After 30 min 20 (4.98 g, 35.0 mmol) was added. The reaction mixture was stirred for 16 h at -60° C before NEt₃ (17.0 g, 168 mmol) was added. The reaction mixture was diluted with Et₂O (200 mL) and washed with water $(2 \times 100 \text{ mL})$ and brine (20 mL). After filtration through silica gel with $Et₂O$ (500 mL) the solvents were removed under reduced pressure. Kugelrohr distillation (120 °C, 20 mbar) yielded 21 (3.56 g, 73 %). ¹H NMR (400 MHz, C₆D₆): δ = 9.17 (s, 1H), 4.86 (t, ⁴J = 1.6 Hz, 1H), 4.81 (m, 1H), 1.75 (tdd, ³J = 7.8, 4 J = 1.2, 1.1 Hz, 2H), 1.27 (qt, 3 J = 7.6, 7.5 Hz, 2H), 0.98 (s, 6H), 0.76 ppm $(t, {}^{3}J=7.3 \text{ Hz}, 3\text{ H}); {}^{13}C \text{ NMR}$ (100 MHz, C₆D₆): $\delta=201.5, 149.5, 111.7,$ 52.1, 46.8, 34.5, 21.8, 20.9, 13.9 ppm; IR (film): $\tilde{v} = 2930, 1740, 1705, 1640,$ 1455, 1370, 1310, 1240, 1190, 1025 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₉H₁₆O: 140.1201; found: 140.1201 $[M]$ ⁺.

4,4-Dimethyl-5-propyl-hexa-2,5-dienoic dimethylamide (22): According to GP 1 21 (701 mg, 5.00 mmol), NaH (156 mg, 6.50 mmol) and the phosphonate (1.79 g, 8.00 mmol) in Et₂O (50 mL) for 16 h. Silica gel chromatography (CH:EE 3:1) gave 22 (946 mg, 90%). $R_f = 0.3$ (CH/EE 3:1); ¹H NMR (300 MHz, C₆D₆): δ = 7.16 (d, ³J = 15.4 Hz, 1H), 6.08 (d, ³J = 15.4 Hz, 1H), 4.99 (dt, $\frac{2}{J}=0.8$, $\frac{4}{J}=0.8$ Hz, 1H), 4.84 (td, $\frac{4}{J}=1.2$, $\frac{2}{J}=$ 1.2 Hz, 1 H), 2.73 (brs, 3 H), 2.38 (brs, 3 H), 1.72 (tdd, $3J = 7.8$, $2J = 1.0$, 1.1 Hz, 2H), 1.23 (qt, $3J=7.6$, 7.5 Hz, 2H), 0.91 (s, 6H), 0.65 ppm (t, $3J=$ 7.3 Hz, 3H); ¹³C NMR (75 MHz, C₆D₆): δ = 166.0, 154.6, 153.5, 117.8, 108.7, 42.5, 36.3, 35.2, 34.3, 26.4, 22.4, 14.2 ppm; IR (film): $\tilde{v} = 2960$, 1660, 1615, 1465, 1390, 1260, 1140, 990, 895, 715, 630 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₃H₂₃NO: 209.1780; found: 209.1786 [M]⁺.

4-Methyl-4-(2-propyl-oxiranyl)-pent-2-enoic dimethylamide (9): According to GP 2 22 (250 mg, 1.19 mmol) and mCPBA (77%, 535 mg, 2.39 mmol) in CH_2Cl_2 (29 mL) for 16 h. Silica gel chromatography (CH/ EE 7:3) gave 9 (246 mg, 92%). $R_f = 0.1$ (CH/EE 1:1); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3): \delta = 6.84 \text{ (d, } ^3J = 15.4 \text{ Hz}, 1 \text{ H}), 6.20 \text{ (d, } ^3J = 15.4 \text{ Hz},$ 1H), 3.05 (s, 3H), 3.29 (s, 3H), 2.67 (d, \overline{J} = 4.3 Hz, 1H), 2.56 (d, \overline{J} = 4.3 Hz, 1H), 1.72 (ddd, $^{2}J=14.6$, $^{3}J=10.7$, 5.9 Hz, 1H), 1.58 (ddd, $^{2}J=$ $14.7, \frac{3}{7} = 10.7, 5.2$ Hz, 1H), 1.25–1.11 (m, 2H), 1.09 (s, 3H), 1.04 (s, 3H), 0.83 ppm (t, ${}^{3}J = 7.3$ Hz, 3H); ¹³C NMR (100 MHz, C₆D₆): $\delta = 165.9$, 151.3, 118.9, 62.2, 47.3, 40.2, 36.4, 35.2, 32.6, 23.6, 22.8, 17.8, 14.5 ppm; IR (film): $\tilde{v} = 2965$, 2875, 1660, 1620, 1470, 1395, 1280, 1145, 990, 945, 850, 820, 715, 630 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₃H₂₃NO₂: 225.1729; found: 225.1731 [M]⁺.

2-(2-Hydroxymethyl-3,3-dimethyl-2-propyl-cyclopropyl)-N,N-dimethyl-

acetamide (10): According to GP 3 9 (169 mg, 0.75 mmol), $[Cp_2TiCl_2]$ (28.0 mg, 113 μ mol), collidine hydrochloride (296 mg, 1.88 mmol) and zinc dust (98.0 mg, 1.50 mmol) for 72 h. Silica gel chromatography (CH/ EE 1:1) gave 10 (162 mg, 95%, dr 63:37). $R_f = 0.4$ (EE); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3): \delta = 3.83 \text{ (d, }^2J = 10.7 \text{ Hz}, 1 \text{ H}), 3.69* \text{ (d, }^2J = 11.7 \text{ Hz},$ 1H), 3.48* (dd, $^2J=12.1$, $^3J=1.1$ Hz, 1H), 3.31 (dd, $^2J=10.6$, $^3J=1.3$ Hz, 1H), 3.31* (s, 3H), 3.27 (s, 3H), 2.94* (s, 3H), 2.93 (s, 3H), 2.81* (dd, $^{2}J=15.4$, $^{3}J=3.5$ Hz, 1H), 2.56 (dd, $^{2}J=17.3$, $^{3}J=4.2$ Hz, 1H), 2.11* (ddd, $^{2}J=14.1, {}^{3}J=11.6, 4.9$ Hz, 3H), 2.02* (d, $^{3}J=10.9, 1$ H), 1.98 (dd, $^{2}J=$ 15.3, ${}^{3}J=11.2$ Hz, 1H), 1.68 (ddd, ${}^{2}J=13.9$, ${}^{3}J=11.5$, 4.7 Hz, 1H), 1.60-1.46* (m, 1H), 1.60–1.46* (m, 1H), 1.35–1.15* (m, 1H), 1.35–1.15 (m, 1H), 1.11 (s, 3H), 1.10* (s, 3H), 1.01 (s, 3H), 0.97* (s, 3H), 0.94 (dddd, $^{2}J=14.1, {}^{3}J=11.7, 4.8, 1.0$ Hz, 1H), 0.86 (t, $^{3}J=7.3$ Hz, 3H), 0.86* (t, $^{3}J=$ 7.3 Hz, 3H), 0.83 (dddd, $\frac{2}{J}$ =14.0, $\frac{3}{J}$ =11.5, 4.8, 1.7 Hz, 1H), 0.72 (dd, $3J=10.9$, 4.2 Hz, 1H), 0.41* ppm (dd, $3J=11.6$, 3.5 Hz, 1H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3): \delta = 174.6^*, 173.3, 65.6, 60.3^*, 37.9^*, 36.9, 35.8^*, 35.6,$ 34.5, 32.1, 31.9*, 30.4*, 29.1*, 28.8, 28.1*, 27.7, 23.8*, 23.7, 22.9, 21.9*, 19.9*, 19.5, 17.8, 16.8*, 14.6, 14.5* ppm; IR (film): $\tilde{v} = 3395$, 2950, 1625, 1400, 1260, 1130, 1015, 795 cm⁻¹; HRMS (EI, 70 eV): m/z: calcd for $C_{13}H_{25}NO_2$: 227.1885; found: 227.1877 [M]⁺.

N,N-Dimethyl-3-(1-vinyl-cyclohexyl)-acrylamide (24): According to GP 1 the phosphonate $(2.43 \text{ g}, 10.9 \text{ mmol})$, NaH $(0.23 \text{ g}, 9.10 \text{ mmol})$ and 23 (1.00 g, 7.25 mmol) in Et₂O (40 mL) for 15 h. Silica gel chromatography (CH/EE 98:2) gave 24 (1.08 g, 48%). $R_f = 0.2$ (CH/EE 96:4); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3)$: $\delta = 6.7 \text{ (dd, }^3 J = 15.4, \frac{4J}{3} = 2.8 \text{ Hz}, 1 \text{ H}), 6.11 \text{ (dd, }^3 J =$ 15.4, $5J=2.3$ Hz, 1H), 5.64 (ddd, $3J=17.6$, $3J=10.8$, $4J=2.8$, $5J=2.3$ Hz, 1H), 5.05 (dddd, $3J=10.8$, $5J=1.2$ Hz, 1H), 4.95 (dddd, $3J=17.7$, $5J=$ 1.3 Hz, 1H), 3.00 (s, 3H), 2.94 (s, 3H), 2.20–2.0 (m, 2H), 1.67–1.31 ppm (m, 8H); ¹³C NMR (100 MHz, CDCl₃): $\delta = 167.2, 152.7, 144.4, 118.2,$ 113.8, 43.2, 37.4, 35.8, 35.4, 26.1, 22.2 ppm; IR (neat): $\tilde{v} = 3485$, 2930, 2855, 1650, 1450, 1390, 1264 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for $C_{13}H_{21}NO$: 207.1623, found 207.1620 [M]⁺.

N,N-Dimethyl-3-(1-oxiranyl-cyclohexyl)-acrylamide (11): According GP 2 24 (1.00 g, 4.80 mmol) and mCPBA (1.25 g (77%), 7.24 mmol) in CH2Cl2 (40 mL) for 16 h. Silica gel chromatography (CH/EE 60:40) yield to **11** (0.72 g, 67%). $R_f = 0.2$ (CH/EE 60:40); ¹H NMR (400 MHz, CDCl₃): $\delta = 6.62$ (d, $\frac{3}{J} = 15.8$ Hz, 1H), 6.26 (d, $\frac{3}{J} = 15.8$ Hz, 1H), 3.04 (s, 3H), 2.96 (s, 3H), 2.76 (dd, $3J=3.9$, $3J=2.9$ Hz, 1H), 2.76 (dd, $2J=4.7$, $3J=3.9$ Hz, 1H), 2.55 (dd, $4J=4.7$, $3J=3.9$ Hz, 1H), 1.71–1.65 (m, 1H), 1.65–1.50 (m, 5H), 1.48–1.30 (m, 3H), 1.27–1.15 ppm (m, 1H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3): \delta = 166.7, 147.8, 121.4, 58.5, 39.9, 37.5, 35.8, 33.3, 31.7,$ 26.0, 22.0, 21.9 ppm; IR (film): 3445, 2925, 1655, 1610, 1395, 1150 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₃H₂₁NO₂: 223.1572, found 223.1575 $[M]$ ⁺.

2-Hydroxymethyl-spiro[2.5]octane-1-carboxylic dimethylamide (12): According to GP 3 11 (223 mg, 1.00 mmol), $[Cp_2TiCl_2]$ (25 mg, 0.10 mmol), collidine hydrochloride (394 mg, 2.50 mmol) and zinc dust (131 mg, 2.00 mmol) for 16 h. Silica gel chromatography (CH/EE $90:10 \rightarrow 50:50$) gave 12 (202 mg, 90%). $R_f = 0.1$ (CH/EE 90:10); ¹H NMR (400 MHz, CDCl₃): $\delta = 6.62$ (d, $\delta J = 15.8$ Hz, 1 H), 6.26 (d, $\delta J = 15.8$ Hz, 1 H), 3.04 (s, 3H), 2.96 (s, 3H), 2.76 (dd, $3J=3.9$, $3J=2.9$ Hz, 1H), 2.76 (dd, $2J=4.7$, $3J=3.9$ Hz, 1H), 2.55 (dd, $4J=4.7$, $3J=3.9$ Hz, 1H), 1.71–1.65 (m, 1H), 1.65–1.50 (m, 5H), 1.48–1.30 (m, 5H), 1.27–1.15 ppm (m, 1H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: $\delta = 174.6^*$, 173.6, 63.9^{*}, 62.5, 58.8, 58.6, 39.7^{*}, 39.3^{*}, 37.9*, 36.9, 35.9*, 35.8, 33.2, 31.9, 28.8*, 28.0, 26.9*, 26.7, 26.3*, 26.2, 26.0, 25.7*, 25.5, 23.6* ppm; IR (film): $\tilde{v} = 3400$, 2925, 1725, 1635, 1400, 1265, 1025 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₃H₂₃NO₂: 225.1729, found 225.1723 [M]⁺.

4-Methyl-1-oxa-spiro[2.5]octane-4-carbaldehyde (26): Compound 25 (4.69 g, 30.0 mmol) was added to a solution of Celite (11.0 g), 4 Å molecular sieves (1.78 g) and PCC (7.54 g, 35 mmol) in CH_2Cl_2 (125 mL) and stirred for 5 h at room temperature. After filtration through Florisil the solvent was removed under reduced pressure and the crude product purified by silica gel chromatography (CH/EE 47:3) to yield 26 (2.83 g, 61 %). $R_{\rm f}$ = 0.3 (CH/EE 9:1); ¹H NMR (400 MHz, C₆D₆): δ = 9.50 (s, 1H), 2.27 (dd, \degree J = 4.2, \degree J = 0.6 Hz, 1H), 1.94 (d, \degree J = 4.2 Hz, 1H), 1.90 (ddd, $^{2}J=13.2$, $^{3}J=8.7$, 4.5 Hz, 1H), 1.52–1.42 (m, 1H), 1.34–1.26 (m, 2H), 1.25–1.14 (m, 3H), 0.94 (ddd, $\frac{2}{J}$ =13.6, $\frac{3}{J}$ =7.7, 4.1 Hz, 1H), 0.74 ppm (s, 3H); ¹³C NMR (75 MHz, C₆D₆): δ = 203.9, 60.2, 49.1, 48.5, 32.2, 31.9, 24.5, 21.2, 16.9 ppm; IR (film): $\tilde{v} = 2930$, 1725, 1450, 1040, 775 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₉H₁₄O₂: 154.0994; found: 154.1001 $[M]$ ⁺.

N,N-Dimethyl-3-(4-methyl-1-oxa-spiro[2.5]oct-4-yl)-acrylamide (13): According to GP 1 26 (540 mg, 3.50 mmol), NaH (96.0 mg, 4.00 mmol) and the phosphonate (1.00 g, 4.50 mmol) in THF (35 mL) for 2 h. Kugelrohr distillation (230 $^{\circ}$ C, 1 mbar) and silica gel chromatography (CH/EE 63:37) gave 13 (580 mg, 74%). $R_f = 0.4$ (EE); ¹H NMR (400 MHz, C₆D₆): δ = 7.17 (d, ²J = 15.7 Hz, 1H), 6.43 (d, ²J = 15.7 Hz, 1H), 2.72 (brs, 3H), 2.44 (dd, $^2J=4.7$, $^4J=1.2$ Hz, 1H), 2.39 (brs, 3H), 1.99 (d, $^2J=4.8$ Hz, 1H), AB signal ($\delta_{\text{Al}}=1.68$, $\delta_{\text{Bl}}=1.65$, $J_{\text{AB}}=13.3$ Hz, additionally split by $3J=6.7, 3.9, 4J=1.2$ Hz, 2H), 1.63–1.51 (m, 1H), 1.40–1.19 (m, 4H), 1.12 (ddd, $\frac{2}{J}=13.4$, $\frac{3}{J}=9.4$, 4.0 Hz, 1H), 0.83 ppm (s, 3H); ¹³C NMR $(100 \text{ MHz}, \text{ C}_6\text{D}_6)$: $\delta = 166.1, 150.2, 120.9, 62.0, 49.4, 40.3, 38.4, 36.4, 35.1,$ 32.0, 25.5, 22.5, 22.3 ppm; IR (film): $\tilde{v} = 2935$, 2860, 1660, 1620, 1495, 1395, 1265, 1145, 1035, 990, 925, 900, 835, 800, 765, 710, 620 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₃H₂₁NO₂: 223.1572; found: 223.1570 [M]⁺.

2-(1-Hydroxymethyl-6-methyl-bicyclo[4.1.0]hept-7-yl)-N,N-dimethylacetamide (14): According to GP 3 13 (223 mg, 1.00 mmol), $[Cp_2Tic1_2]$ $(24.9 \text{ mg}, 100 \text{ µmol})$, collidine hydrochloride $(394 \text{ mg}, 2.50 \text{ mmol})$ and zinc dust (131 mg, 2.00 mmol) for 16 h. Silica gel chromatography (CH/ EE 1:1) gave **14** (162 mg, 72%). $R_f = 0.2$ (EE); ¹H NMR (300 MHz, C_6D_6 : $\delta = 4.27^*$ (brs, 1H), 4.06 (brs, 1H), 3.90* (d, ²J = 10.2 Hz, 1H), 3.76 (s, 2H), 3.61* (d, $^2J=12.2$ Hz, 1H), 2.58* (s, 3H), 2.56 (s, 3H), 2.36 (dd, \degree J = 15.4, \degree J = 3.6 Hz, 1H), 2.24–2.10* (m, 1H), 2.19 (s, 3H), 2.18– 2.12^* (dm, $3J=4.3$ Hz, 1H), 2.16^* (s, 3H), 2.01 (ddd, $3J=14.2$, $3J=9.3$ Hz, 5.2 Hz, 1H), 1.84* (dd, ²J = 16.9, ³J = 11.0 Hz, 1H), 1.83 (dd, ²J = 15.3, ³J = 11.5 Hz, 1H), 1.64 (dddd, $\mathcal{Y}=13.4$, $\mathcal{Y}=4.8$, 4.8, 1.5 Hz, 1H), 1.47–0.92 $(m, 6H), 1.47-0.92*$ $(m, 7H), 1.06*$ (s, 3H), 0.89 (s, 3H), 0.76* (dd, $3J=$ 10.8, 4.2 Hz, 1H), 0.69 ppm (dd, $3J=11.5$, 3.6 Hz, 1H); ¹³C NMR $(75 \text{ MHz}, \text{ C}_6\text{D}_6)$: $\delta = 174.0, 172.3^*, 70.0^*, 64.2, 36.8, 35.2^*, 35.2, 35.0^*,$

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33.7*, 29.7, 29.4*, 29.9, 28.8*, 28.6*, 28.4, 28.4, 27.3, 24.3*, 24.2*, 22.7*, 22.6*, 22.3, 21.8, 21.8*, 21.6, 17.2 ppm; IR (film): $\tilde{v} = 3390$, 2925, 1720, 1635, 1400, 1260, 1145, 1010, 615 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for $C_{13}H_{23}NO_2$: 225.1729; found: 225.1725 $[M]$ ⁺.

1-Methyl-2-methylene-cycloheptanecarboxylic ethyl ester (28): NaH $(1.78 \text{ g}, 70.5 \text{ mmol})$ and DMSO (40 mL) were heated for 1 h at 70 °C. To the stirred solution methyltriphenylphosphonium bromide (25.5 g, 70.5 mmol) in DMSO (80 mL) was added. After 1 h 27 (13.3 g, 67.2 mmol) was added and the reaction mixture stirred for 16 h at 55° C. The reaction was interrupted by adding water (80 mL) and extracting with cyclohexane $(3 \times 50 \text{ mL})$. The combined organic layers were washed with water/DMSO (1:1, 80 mL), with brine (50 mL) and dried (MgSO₄). The solvents were removed under reduced pressure and the crude product purified by silica gel chromatography (CH/EE 95:5) to yield 28 (9.30 g, 70%). $R_f = 0.2$ (CH/EE 95:5); ¹H NMR (400 MHz, CDCl₃): $\delta =$ 4.94 (s, 1H), 4.85 (s, 1H), 4.10 (t, $3I = 7.1$ Hz, 2H), 2.26 (m, 2H), 2.15 (d, $3J=9.2$ Hz, 1H), 2.12 (d, $3J=9.3$ Hz, 1H), 1.75–1.63 (m, 3H), 1.61–1.50 $(m, 2H)$, 1.49–1.32 $(m, 3H)$, 1.31 $(s, 1H)$, 1.20 ppm $(t, \frac{3}{5}J = 7.1 \text{ Hz}, 3H)$; ¹³C NMR (100 MHz, CDCl₃): δ = 176.9, 113.0, 60.6, 50.9, 38.4, 35.1, 31.4, 30.5, 25.0, 24.5, 14.2 ppm; IR (film): $\tilde{v} = 2925$, 1725, 1630, 1445, 1235, 1155, 1105 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₇H₂₀O₂: 196.1463; found 196.1471 [M]⁺.

(1-Methyl-2-methylene-cycloheptyl)-methanol (29): Compound 28 (9.81 g, 50.0 mmol) was added dropwise to a suspension of $LiAlH₄$ (0.95 g, 25.0 mmol) in dry diethyl ether (100 mL) at 0° C. The reaction mixture was stirred for 10 h at room temperature. Water (1.0 mL), NaOH (1.0 mL, 15% w/v) and water (3.0 mL) were added subsequently. The solid was filtered off and the solvents were removed under reduced pressure. Distillation (b.p. 4.8 mbar/130–135 \degree C) of the crude product lead to **29** (6.32 g, 41 mmol). ¹H NMR (400 MHz, CDCl₃): $\delta = 4.94$ (s, 1H), 4.78 (s, 1H), 3.22 (dd, $\alpha J = 10.6$, $\beta J = 3.6$ Hz, 1H), 3.18 (dd, $\alpha J = 10.4$, $\beta J =$ 9.4 Hz, 1H), 2.16 (dd, $\mu^2 J = 13.1$, $\mu^3 J = 7.6$ Hz, 1H), 1.98 (dd, $\mu^2 J = 13.0$, $\mu^3 J =$ 13.0 Hz, 1H), 1.83–1.77 (m, 1H), 1.70–1.53 (m, 3H), 1.42–1.37 (m, 2H), 1.29–1.18 (m, 4H), 1.03 ppm (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ = 156.1, 113.0, 70.2, 45.1, 37.3, 34.5, 32.2, 31.0, 23.7, 23.2 ppm; IR (film): $\tilde{v} =$ 3385, 2920, 1725, 1630, 1445, 1045 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for $C_{10}H_{18}O$: 154.1358; found 154.1352 [*M*]⁺.

1-Methyl-2-methylene-cycloheptanecarbaldehyd (30): DMSO (9.38 g, 120 mmol) in CH_2Cl_2 (10 mL) was added to a solution of oxalyl chloride (7.62 g, 60.0 mmol) in CH₂Cl₂ (150 mL) at -60° C. After 15 min 29 (7.70 g, 50.0 mmol) was added. The reaction stirred overnight before $NEt₂$ (34.7 mL, 250 mmol) was added. The reaction mixture was diluted with Et₂O (100 mL) and washed with water $(2 \times 100 \text{ mL})$ and brine (50 mL) . Drying $(MgSO_4)$, evaporation and purifying of the crude product by silica gel chromatography (CH/EE 99:1) yield to 30 (5.16 g, 68%). $R_f=0.2$ (CH/EE 98:2); ¹H NMR (400 MHz, CDCl₃): $\delta = 9.40$ (s, 1H), 5.09 (s, 1H), 4.81 (s, 1H), 2.26 (dd, $\mathcal{Y} = 13.1$, $\mathcal{Y} = 1.6$ Hz, 1H), 2.24 (dd, $^{2}J=13.1$, $^{3}J=1.6$ Hz, 1H), 2.15–2.08 (m, 2H), 2.04 (d, $^{2}J=15.1$ Hz, 1H), 1.99 (d, $^2J=14.7$ Hz, 1H), 1.80-1.36 ppm (m, 6H), 1.20 ppm (s, 1H); ¹³C NMR (100 MHz, CDCl₃): δ = 202.6, 151.6, 115.3, 55.2, 35.1, 34.4, 31.6, 24.0, 22.4 ppm; IR (film): $\tilde{v} = 2925$, 2855, 2695, 1725, 1630, 1445, 1370 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₀H₁₆O: 152.1201; found 152.1198 [M]⁺.

N,N-Dimethyl-3-(1-methyl-2-methylen-cycloheptyl)-acryl amide (31): According to GP 1 the phosphonate (3.71 g, 15.0 mmol), NaH (0.33 g, 12.5 mmol) and 30 (1.52 mmol, 10.0 mmol) in Et₂O (40 mL) for 16 h. Silica gel chromatography (CH/EE 50:50) yield to 31 (1.94 g, 93%). R_f = 0.2 (CH/EE 50:50); ¹H NMR (400 MHz, CDCl₃): δ = 6.86 (d, ³J = 15.4 Hz, 1H), 6.09 (d, $3J=15.4$ Hz, 1H), 4.86 (s, 1H), 4.73 (s, 1H), 3.00 (s, 3H), 2.95 (s, 3H), 2.20–2.04 (m, 2H), 1.68–1.30 (m, 8H), 1.14 ppm (s, 1H); ¹³C NMR (100 MHz, CDCl₃): δ = 167.3, 156.3, 153.8, 116.2, 112.9, 45.4, 37.3, 35.7, 34.1, 31.8, 30.7, 29.0, 26.9, 25.4, 23.8 ppm; IR (film): $\tilde{v} = 2930$, 2860, 1710, 1650, 1460, 1370, 1160 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for $C_{13}H_{21}NO$: 221.1780; found 221.1776 $[M]$ ⁺.

N,N-Dimethyl-3-(4-methyl-1-oxa-spiro[2.6]-non-4-yl)-acrylamide (15): According GP 2 31 (1.60 g, 7.70 mmol) and *mCPBA* (1.98 g (77 %), 11.5 mmol) in CH_2Cl_2 (40 mL) for 8 h. Silica gel chromatography (CH/ EE 50:50) yield to **15** (1.56 g, 86%). $R_f = 0.15$ (CH/EE 50:50); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3)$: $\delta = 6.79^*$ (d, $\delta J = 15.6 \text{ Hz}, 1 \text{ H}$), 6.75 (d, $\delta J = 15.4 \text{ Hz}$, 1H), 6.14* (d, $3J=15.6$ Hz, 1H), 6.07 (d, $3J=15.4$ Hz, 1H), 3.01 (s, 6H),

2.93 (s, 6H), 2.59 (dd, $\frac{2J}{-4.9}$, $\frac{4J}{-0.8}$ Hz, 1H), 2.58* (dd, $\frac{2J}{-5.2}$, $\frac{4J}{-}$ 0.8 Hz, 1H), 2.38 (d, $\frac{3}{5}J=4.8$ Hz, 1H), 2.35* (d, $\frac{2}{5}J=5.2$ Hz, 1H), 1.74– 1.40 (m, 20H), 0.94 ppm (s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ = 167.5*, 166.9, 150.9, 150.0*, 118.1*, 117.0, 63.8, 63.0, 52.6, 51.1*, 42.6*, 42.6, 40.0*, 39.8, 37.5*, 37.3, 37.7, 35.6*, 32.7, 32.5*, 30.5, 30.0*, 25.6, 25.2*, 23.7*, 23.1, 23.0*, 21.5 ppm; IR (film): $\tilde{v} = 2925$, 1660, 1615, 1460, 1390, 1270, 1140 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₄H₂₃NO₂: 237.1729, found: 237.1725 $[M]^+$; elemental analysis calcd $(\%)$ for $C_{14}H_{23}NO_2$: C 70.85, H 9.77, N 5.90; found: C 70.16, H 9.42, N 5.86.

2-(1-Hydroxymethyl-7-methyl-bicyclo[5.1.0]oct-8-yl)-N,N-dimethylacetamide (16): According to GP 3 15 (223 mg, 0.94 mmol), $[Cp_2TiCl_2]$

(24 mg, 0.1 mmol), collidine hydrochloride (370 mg, 2.35 mmol) and zinc dust (123 mg, 1.88 mmol) for 16 h. Silica gel chromatography (CH/EE 85:15 \rightarrow 0:100) yield to 16 (200 mg, 88%). R_f =0.05 (CH/EE 85:15). ¹H NMR (300 MHz, CDCl₃): δ = 3.87* (d, ²J = 10.6 Hz, 1H), 3.74 (d, ²J = 12.1 Hz, 1 H), 3.46 (dd, $\mathcal{Y} = 12.2$, $\mathcal{Y} = 1.4$ Hz, 1 H), 3.28* (dd, $\mathcal{Y} = 10.5$, $\mathcal{Y} = 12.2$ 1.8 Hz, 1H), 3.06* (s, 3H), 2.98* (s, 3H), 2.96 (s, 3H), 2.92 (s, 3H), 2.90* $(dd, {}^{3}J=15.3, {}^{5}J=3.0 \text{ Hz}, 2 \text{ H}), 2.65 \text{ (m, 2 H)}, 2.19-1.55 \text{ (m, 11 H)}, 2.19-$ 1.55* (m, 11H), 1.13* (s, 3H), 0.97 (s, 3H), 0.95–0.73 (m, 1H), 0.63* ppm (dd, $3J=12.1$, $5J=2.8$ Hz, 1H); $13C NMR$ (75 MHz, CDCl₃): $\delta = 174.9$, 173.5*, 66.4*, 60.5, 41.2*, 38.0, 37.0*, 36.0, 35.8*, 34.7, 34.3*, 33.5*, 33.4, 32.0, 29.8*, 28.8, 26.7, 26.5*, 26.3*, 22.1*, 15.0 ppm; IR (film): $\tilde{v} = 3340$, 2920, 1620, 1465, 1400, 1020 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for $C_{14}H_{25}NO_2$: 239.1885, found: 239.1889 [M]⁺; elemental analysis calcd (%) for C₁₄H₂₅NO₂: C 70.25, H 10.53, N 5.85; found: C 70.09, H 10.34, N 5.68.

5,5,6-Trimethyl-hepta-2,6-dienoic tert-butyl ester (37): According to GP 1 the phosphonate (379 mg, 3.00 mmol), NaH (96.0 mg, 4.00 mmol) and 36 $(1.26 \text{ g}, 5.00 \text{ mmol})$ in Et₂O (50 mL) for 16 h. Silica gel chromatography (CH/EE 97:3) gave 37 (542 mg, 81%). $R_f = 0.4$ (CH/EE 19:1); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3)$: $\delta = 6.67 \text{ (dt, }^3 J = 15.3, ^3 J = 7.6 \text{ Hz}, 1 \text{ H}), 5.61 \text{ (dt, }^3 J =$ 15.5, $^{4}J=1.4$ Hz, 1H), 4.67 (dq, $^{3}J=1.4$, $^{4}J=1.4$ Hz, 1H), 4.62 (dd, $^{3}J=$ 1.3, $^{4}J=0.5$ Hz, 1H), 2.12 (dd, $^{3}J=7.6$, $^{4}J=1.4$ Hz, 2H), 1.63 (dd, $^{4}J=1.3$, $^{4}J=0.5$ Hz, 3H), 1.37 (s, 9H), 0.96 ppm (s, 6H); 13 C NMR: $\delta=165.9$, 150.9, 145.2, 124.7, 110.2, 80.0, 43.3, 39.0, 28.2, 27.0, 19.4 ppm; IR (film): $\tilde{v} = 2970, 1715, 1650, 1455, 1367, 1335, 1160, 985, 895$ cm⁻¹; HRMS: m/z : calcd for $C_{10}H_{16}O_2$: 168.1150; found: 168.1148 $[M-C_4H_8]^+$.

5-Methyl-5-(2-methyl-oxiranyl)-hex-2-enoic tert-butylester (32): According to GP 2 37 (449 mg, 2.00 mmol) and m-CPBA (672 mg (77%), 3.00 mmol) in CH₂Cl₂ (25 mL) for 1 h. Silica gel chromatography (CH $/$ EE 9:1) gave 32 (387 mg, 81%). $R_f = 0.2$; ¹H NMR (400 MHz, C₆D₆): $\delta =$ 7.06 (dt, $3J=15.5$, $3J=7$, 7 Hz, 1H), 5.84 (dt, $3J=15.5$, $4J=1.4$ Hz, 1H), 2.39 (dq, $3J=4.6$, $4J=0.8$ Hz, 1H), 1.99 (d, $3J=4.5$ Hz, 1H), 1.92, 1.88, AB signal, $^{2}J=14.1$ Hz, additionally split by $^{3}J=7.7$ Hz and $^{4}J=1.4$ Hz, 2H), 1.43 (s, 9H), 0.99 (d. ⁴J=0.8 Hz, 3H), 0.74 (s, 3H), 0.66 ppm (s, 3H); ¹³C NMR (100 MHz, C₆D₆): δ = 165.4, 144.6, 125.8, 79.6, 60.3, 50.7, 41.8, 36.8, 28.2, 23.5, 22.8, 18.4 ppm; IR (film): $\tilde{v} = 2975$, 1715, 1650, 1455, 1370, 1340, 1285, 1165, 1075, 985, 890, 855, 825, 780 cm⁻¹; HRMS: m/z: calcd for $C_{14}H_{24}O_3$: 240.1725; found: 240.1730 $[M]$ ⁺.

2-(2-Hydroxymethyl-2,3,3-trimethyl-cyclobutyl)-acetic tert-butyl ester (33): According to GP 3 32 (240 mg, 1.00 mmol), $[Cp_2Ticl_2]$ (24.9 mg, 100 mg), collidine hydrochloride (394 mg, 2.50 mmol) and zinc dust (131 mg, 2.00 mmol) for 16 h. Silica gel chromatography (CH/EE 86:14) gave 33 (228 mg, 94%, dr 63:37). $R_f = 0.3$ (CH/EE 3:1); ¹H NMR $(400 \text{ MHz}, \text{ C}_6\text{D}_6)$: $\delta = 3.65^*$ (d, $^2J = 10.9 \text{ Hz}, 1 \text{ H}$), 3.60 (d, $^2J = 10.7 \text{ Hz}$, 1 H), 3.41* (d, $\frac{2}{J}$ = 11.0 Hz, 1 H), 3.40 (d, $\frac{2}{J}$ = 10.6 Hz, 1 H), 2.47–2.38 (m, 1H), 2.47–2.38* (m, 1H), 2.33 (brs, 1H), 2.21 (dd, $\overline{z}J = 16.3$, $\overline{z}J = 9.9$ Hz, 1 H), 2.18* (dd, $\frac{2}{J}$ = 17.5, $\frac{3}{J}$ = 7.8 Hz, 1 H), 2.02 (dd, $\frac{2}{J}$ = 16.5, $\frac{3}{J}$ = 5.4 Hz, 1 H), 1.79* (brs, 1 H), 1.63* (dd, ²J = 10.8, ³J = 8.0 Hz, 1 H), 1.52 (dd, ²J = 10.4, ${}^{3}J=8.4$ Hz, 1H), 1.42–1.36* (m, 1H), 1.36* (s, 9H), 1.32 (s, 9H), 1.20 (dd, $^{2}J=10.5$, $^{3}J=10.4$ Hz, 1H), 1.22–1.15* (m, 1H), 1.04* (s, 3H), 1.02 (s, 3H), 0.96 (s, 3H), 0.94* (s, 3H), 0.81 (s, 3H), 0.78* ppm (s, 3H); ¹³C NMR (100 MHz, C₆D₆): δ = 173.7, 173.0*,80.1, 79.8*, 68.8, 65.2*, 45.5, 45.3*, 38.3*, 37.5*, 37.4, 37.1*, 36.6, 36.5, 36.4*, 33.6, 28.1*, 28.0, 25.7*, 24.9, 24.2, 23.7*, 19.0*, 14.8 ppm; IR (film): $\tilde{v} = 3445$, 2970, 1730, 1455, 1370, 1260, 1150, 1030, 950, 850, 765 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₄H₂₆O₃: 186.1256; found: 186.1252 $[M - C_4H_8]$ ⁺.

5,5,6-Trimethyl-hepta-2,6-dienoic dimethylamide (38): According to GP 1 36 (505 mg, 4.00 mmol), NaH (120 mg, 5.00 mmol) and 36 (1.34 g, 6.00 mmol) in $Et₂O$ (50 mL) for 16 h. Silica gel chromatography (CH/EE

4:1) gave 38 (631 mg, 81%). $R_f = 0.4$ (CH/EE 1:1); ¹H NMR (400 MHz, CDCl₃): $\delta = 6.67$ (dt, $\delta J = 15.0$, $\delta J = 7$, 5 Hz, 1H), 6.16 (dt, $\delta J = 15.1$, $\delta J = 15.1$ 1.4 Hz, 1H), 4.70 (dq, $3I=1.4$, $4J=1.4$ Hz, 1H), 4.69 (dd, $3J=1.5$, $4J=$ 0.7 Hz, 1H), 3.02 (brs, 3H), 2.95 (brs, 3H), 2.20 (dd, $3J=7.6$, $4J=1.3$ Hz, 2H), 1.71 (dd, $^{4}J=1.4$, $^{4}J=0.56$ Hz, 3H), 1.04 ppm (s, 6H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_2)$: $\delta = 166.7, 151.0, 142.9, 122.2, 110.1, 43.8, 39.0, 37.3,$ 35.5, 27.0, 19.5 ppm; IR (film): $\tilde{v} = 2965$, 1660, 1620, 1450, 1395, 1265, 1145, 1065, 980, 890, 825, 740, 620 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for $C_{12}H_{21}NO_2$: 195.1623; found: 195.1622 $[M]$ ⁺.

5-Methyl-5-(2-methyl-oxiranyl)-hex-2-enoic dimethylamide (34): According to GP 2 38 (391 mg, 2.00 mmol) and mCPBA (77%, 672 mg, 3.00 mmol) in CH_2Cl_2 (25 mL) for 16 h. Silica gel chromatography (CH/ EE 1:1 \rightarrow 0:1) gave 34 (402 mg, 95%). R_f =0.1 (CH/EE 1:1); ¹H NMR $(400 \text{ MHz}, \text{ C}_6\text{D}_6)$: $\delta = 7.11 \text{ (dt, } {}^3J = 14.9, {}^3J = 7, 6 \text{ Hz}, 1 \text{ H}), 6.03 \text{ (dt, } {}^3J =$ 14.9, $^{4}J=1.5$ Hz, 1H), 2.71 (brs, 3H), 2.49 (dq, $^{3}J=4.7$, $^{4}J=0.6$ Hz, 1H), 2.36 (brs, 3H), 2.06 (s, 3H), 1.99 (AB signal, $\mathcal{Y}=13.9$ Hz, additionally split by ${}^{3}J=7.7$ and ${}^{4}J=1.4$ Hz, 2H), 1.07 (d, ${}^{4}J=0.6$ Hz, 1H), 0.83 (s, 3H), 0.76 ppm (s, 3H); ¹³C NMR (100 MHz, C₆D₆): δ = 165.8, 142.1, 123.3, 60.5, 51.0, 42.2, 36.8, 36.4, 35.2, 23.6, 22.8, 18.5 ppm; IR (film): $\tilde{v} =$ 2970, 1660, 1615, 1495, 1395, 1265, 1150, 1060, 980, 890, 815, 775 cm⁻¹; HRMS: m/z : calcd for C₁₂H₂₀NO₂: 210.1494; found: 210.1496 [M-H]⁺.

2-(2-Hydroxymethyl-2,3,3-trimethyl-cyclobutyl)-N,N-dimethylacetamide (35): According to GP 3 34 (211 mg, 1.00 mmol), $[Cp_2TiCl_2]$ (24.9 mg, 100 mg), collidine hydrochloride (394 mg, 2.50 mmol) and zinc dust (131 mg, 2.00 mmol) for 72 h. Silica gel chromatography (EE) gave 35 (200 mg, 94%, dr 72:28). $R_f = 0.3$ (EE); ¹H NMR (400 MHz, C₆D₆): $\delta =$ 4.63 (dd, $\mathrm{^{3}J=10.0, 2.7 \, Hz, 1H}$), 4.45* (d, $\mathrm{^{3}J=6.8 \, Hz, 1H}$), 4.05* (d, $\mathrm{^{2}J=}$ 11.0 Hz, 1 H), 4.01 (dd, $^2J=11.0$, $^3J=1.8$ Hz, 1 H), 3.56* (dd, $^2J=10.7$, $^3J=$ 7.2 Hz, 1 H), 3.43 (dd, $\frac{2}{J}$ =10.7, $\frac{3}{J}$ =10.2 Hz, 1 H), 2.71 (dddd, $\frac{3}{J}$ =11.1, 10.7, 8.4, 3.0 Hz, 1H), 2.55* (s, 3H), 2.50 (s, 3H), 2.37* (dd, $\mathcal{Y}=14.5$, $\mathcal{Y}=$ 10.5 Hz, 1 H), $2.26*$ (dddd, $3J=9.9$, 9.9, 8.4, 4.1 Hz, 1 H), $2.22*$ (s, 3 H), 2.15 (s, 3H), 2.11* (dd, $^2J=14.5$, $^3J=4.0$ Hz, 1H), 2.03 (dd, $^2J=17.3$, $^3J=$ 11.1 Hz, 1 H), 1.77 (dd, $^2J=17.5$, $^3J=2.9$ Hz, 1 H), 1.67* (dd, $^2J=0.8$, $^3J=$ 8.3 Hz, 1 H), 1.57 (dd, $\mathcal{Y} = 10.4$, $\mathcal{Y} = 8.4$ Hz, 1 H), 1.41* (dd, $\mathcal{Y} = 10.2$, $\mathcal{Y} = 10.3$ 10.2 Hz, 1 H), 1.28 (dd, $\frac{2}{J}$ = 10.6, $\frac{3}{J}$ = 10.6 Hz, 1 H), 1.25* (s, 3 H), 1.07 (s, 3H), 1.01 (s, 3H), 1.00* (s, 3H), 0.84 (s, 3H), 0.82* ppm (s, 3H); ¹³C NMR (100 MHz, C₆D₆): δ = 173.4*, 173.1, 68.8, 64.7*, 46.1*, 45.7, 38.6*,37.6*,37.5, 36.8*, 36.6*, 36.4, 36.1, 35.0*, 34.9*, 34.3, 34.1, 25.9*, 24.5, 24.0, 23.3*, 19.3*, 14.8 ppm; IR (film): $\tilde{v} = 3265$, 2935, 1610, 1505, 1560, 1410, 1260, 1160, 1045, 800, 770, 710, 690 cm⁻¹; HRMS (EI, 70 eV): m/z : calcd for C₁₂H₂₃NO₂: 213.1729; found: 213.1729 [M]⁺.

Acknowledgement

We are indebted to the Deutsche Forschungsgemeinschaft (SFB 624) and the Fonds der Chemischen Industrie (Dozentenstipendium to A.G. and Sachbeihilfen) for financial support.

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Received: July 5, 2004 Published online: August 24, 2004